

# Full-scale Mars Science Laboratory Tiled Heatshield Material Response

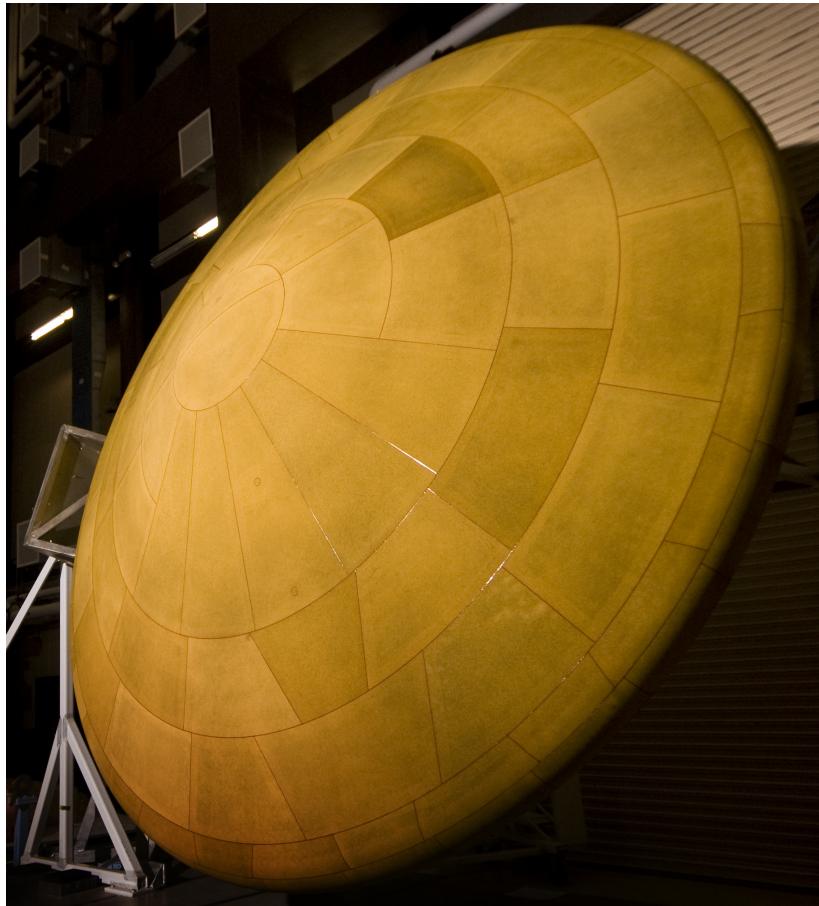
**Jeremie B. E. Meurisse<sup>1</sup>**  
**Jean Lachaud<sup>2</sup>**  
**Chun Y. Tang<sup>2</sup>**  
**Nagi N. Mansour<sup>2</sup>**

<sup>1</sup> Science & Technology Corporation at NASA Ames Research Center, Moffett Field, CA 94035, USA  
<sup>2</sup> NASA Ames Research Center, Moffett Field, CA 94035, USA

**9<sup>th</sup> Ablation Workshop**  
Montana State University, August 30<sup>th</sup> - 31<sup>st</sup>, 2017



# Mars Science Laboratory landed Curiosity in 2012



tiled MSL PICA heatshield  
Lockheed Martin

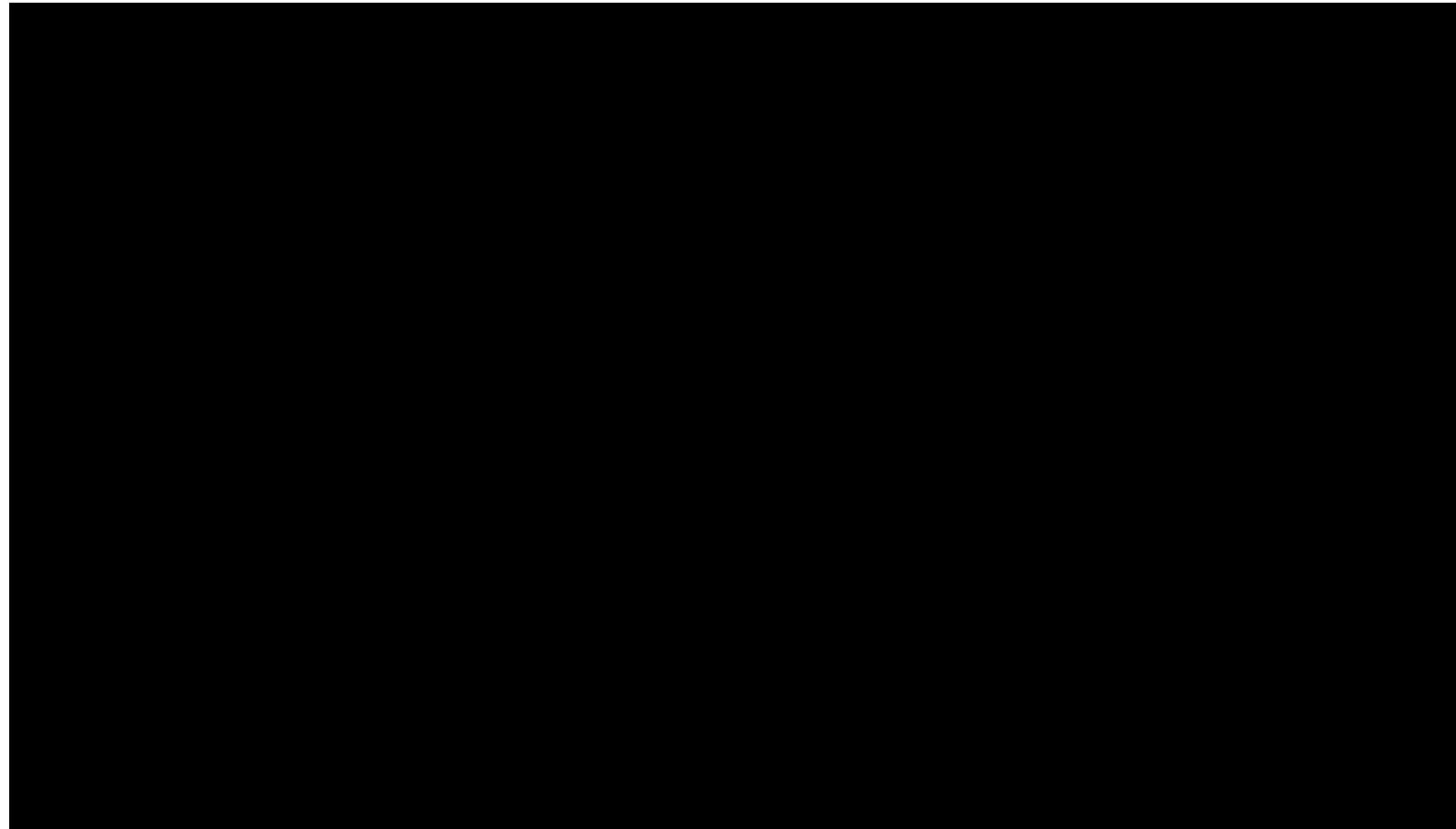
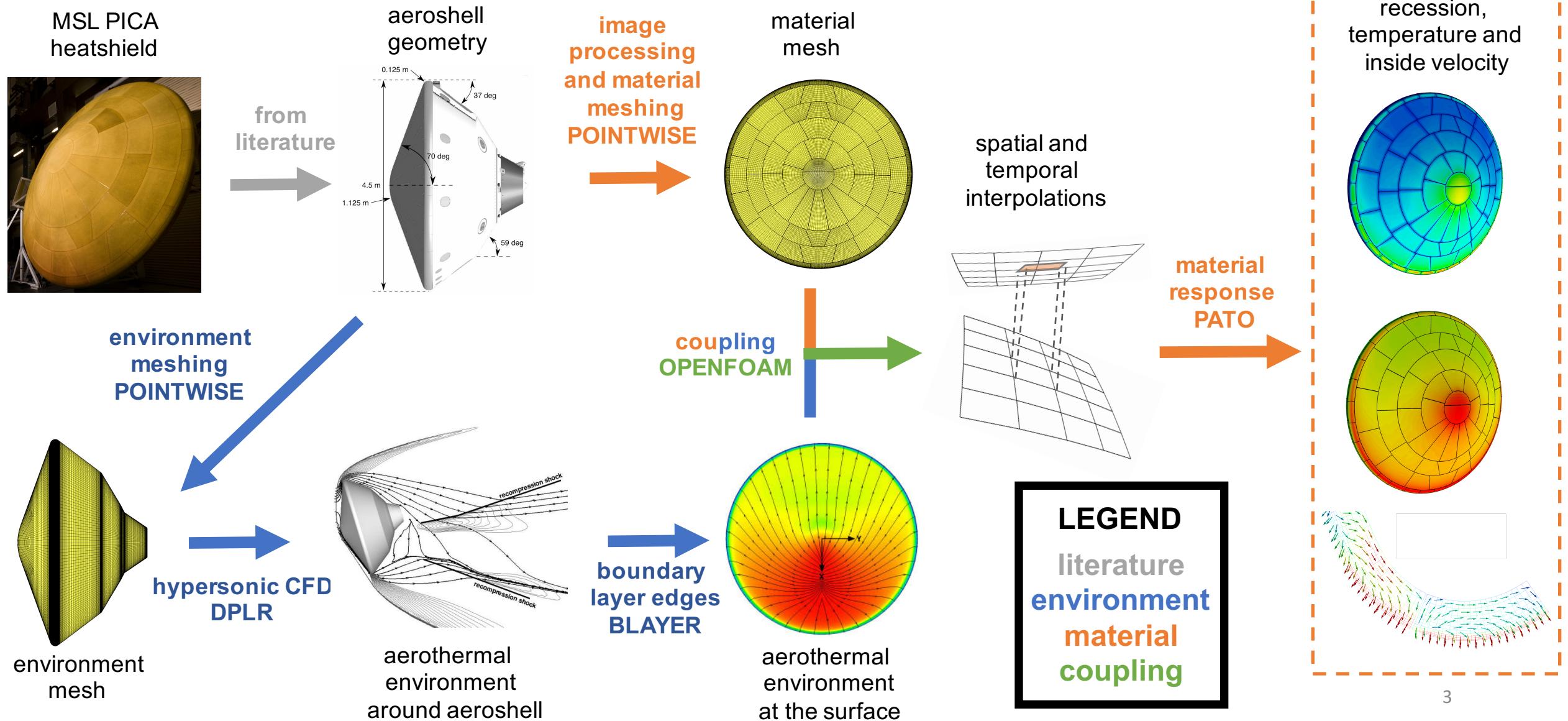
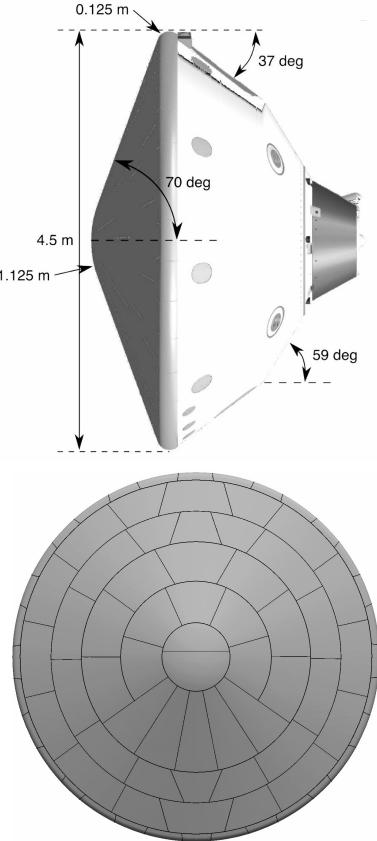


Illustration of MSL Mars entry on August 2012  
credit: NASA JPL

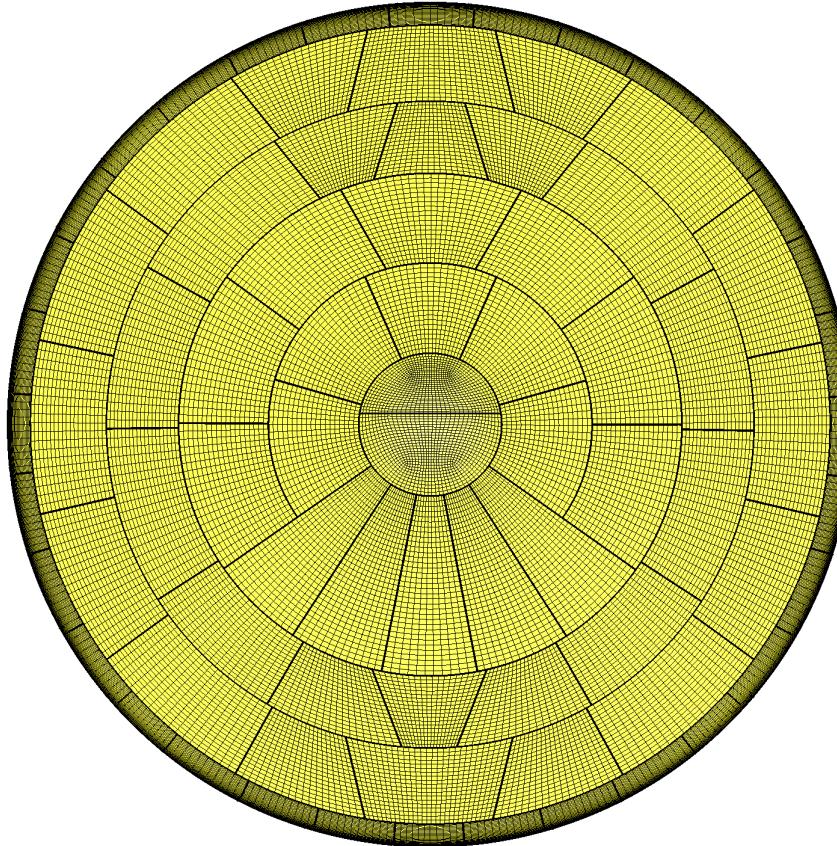
# Overview – coupling aerothermal environment and material response



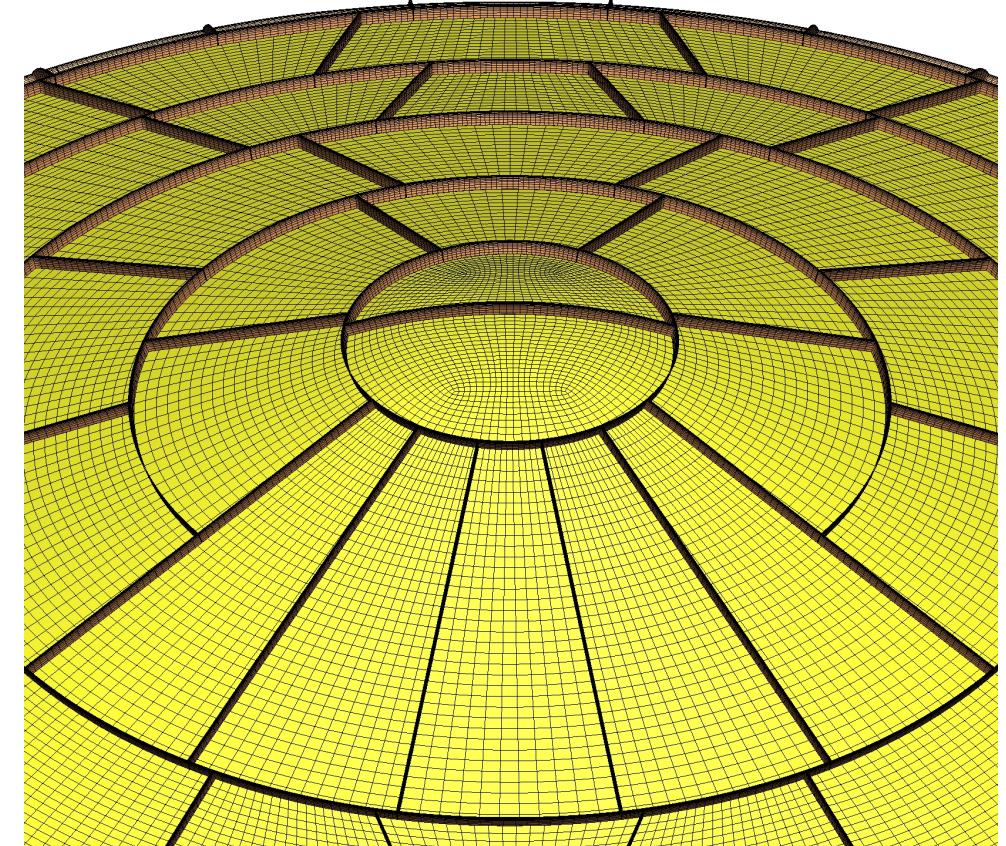
# Computational domain of the entire MSL heatshield



aeroshell geometry with  
113 PICA heatshield tiles



2 million cells mesh  
of the tiled heatshield



heatshield material in 2 regions  
bounding agent + porous tiles

# PATO\* is used for the material response model

## Pyrolysis

$$\partial_t \chi_{i,j} = (1 - \chi_{i,j})^{m_{i,j}} T^{n_{i,j}} A_{i,j} \exp\left(\frac{-E_{i,j}}{RT}\right)$$

$$\Pi = \sum_{i=1}^{N_g} \sum_{j=1}^{N_p} \sum_{k=1}^{P_j} \zeta_{i,j,k} \epsilon_{i,0} \rho_{i,0} F_{i,j} \partial_t \chi_{i,j}$$

## Mass and momentum conservation

$$\partial_t \left( \frac{\epsilon_g M}{RT} p \right) - \partial_x \cdot \left( \frac{pM}{RT} \left[ \frac{1}{\mu} \bar{K} + \frac{1}{p} \bar{\beta} \right] \right) = \Pi$$

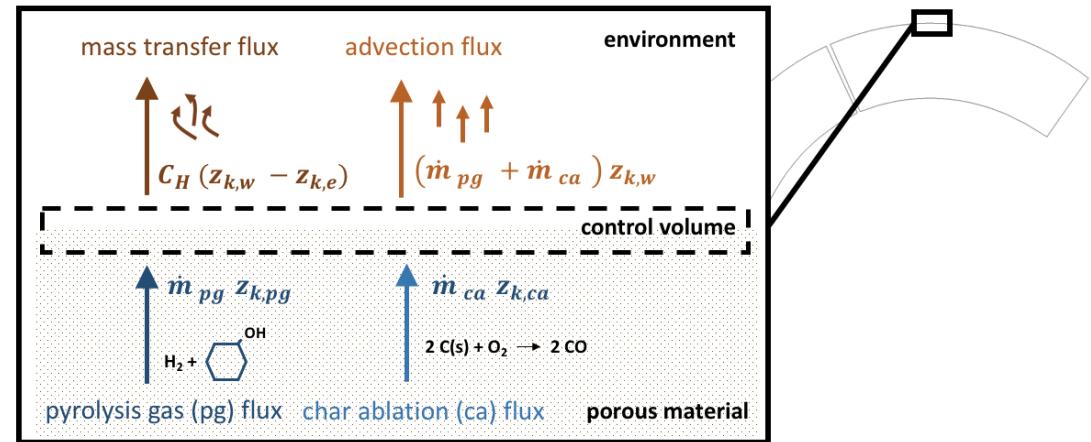
## Energy conservation

$$\sum_{i=1}^{N_p} [(\epsilon_i \rho_i c_{p,i}) \partial_t T] - \partial_x \cdot (\bar{k} \cdot \partial_x T)$$

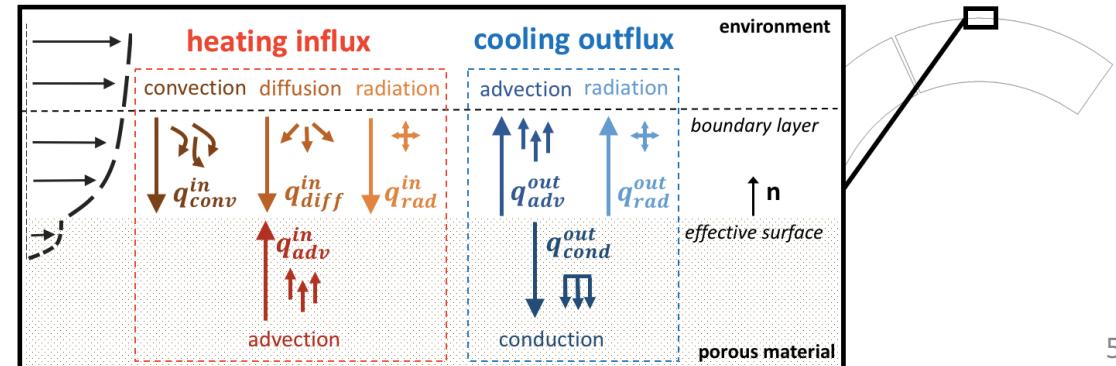
$$= \sum_{i=1}^{N_p} [h_i \partial_t (\epsilon_i \rho_i)] - \partial_t (\epsilon_g \rho_g h_g - \epsilon_g p) + \partial_x \cdot (\epsilon_g \rho_g h_g v_g)$$

\* PATO = Porous-material Analysis Toolbox based on OpenFOAM

**Surface mass balance**  $\longrightarrow h_w \dot{m}_{ca}$



**Surface energy balance**  $\longrightarrow T_w$



# A new GUI implemented in PATO

PATO (Porous-material Analysis Toolbox based on OpenFOAM)

**1. Set Case**

**case definition**

boundaryLayerApprox

failureFraction

physicsBasedErosionModel

model# (Press enter)

volAblation

constrainedEquilibrium

wallSpeciesDiffusion

axiSymmetricalFlux

fluxFactorNormal

fluxFactorCenter

fluxFactorProjection

**2. Environment**

finiteRate

multiComponent

elementConservation

mutationTT

mutationBprime

detailedSolidEnthalpies

readBoundaryTecplotTable

**constant/setCase**

```
constrainedEquilibrium yes; // reaction-limited surface oxidation rate - only with mutationBprime ('on-the-fly')
// yes: finite-rate, available C+O->CO (using rate of Park, bounded by equilibrium)
// no: equilibrium

wallSpeciesDiffusion 1; // this coefficient can take any number; diffusion enthalpy flux = - rhoeUeCh * h_w * wallSpeciesDiffusion
// 1 : State-of-the-art approximation for species diffusion
// 0 : Diffusion neglected

// 1-b) Options for axi-symmetrical heat fluxes - radial evolution factor read from file 'constant/environmentProperties/fluxFactorMap'
axiSymmetricalFlux no;
// "no" for uniform heat load (value inputed in 'constant/environmentProperties/boundaryConditions' is used uniformly)
// "yes" for axiSymmetrical heat load, the three following fluxFactors are used:
// 1- choose normal to directions along which the flux varies (e.g. axis of an IsoQ sample for example). Use X= (1 0 0), Y= (0 1 0), or Z= (0 0 1):
fluxFactorNormal (1 0 0);
// 2- choose center of axi-symmetry
fluxFactorCenter (0.0 0.0 0.0);
// 3- choose type of projection
fluxFactorProjection no;
// "yes": projection on the plane perpendicular to the Normal and passing by the center (recommended for plane surfaces: e.g. top of a cylinder);
// "no": projection directly on the Normal itself; that is, the distance from the center is directly computed by projection on the Normal (recommended for convex surfaces, e.g. IsoQ samples).
// **** End Options for axisymmetrical heat fluxes

// Step 2 - Set in-depth model (all no = type 2 base model)
finiteRate no;
// no: Equilibrium chemistry is used based on the elemental fractions of the pyrolysis gases.
// yes: Finite-rate chemistry based on species fractions (species production in the pyrolysis reactions, finite-rate chemistry mechanism)
// The chemistry mechanisms and their chemistry databases are selected in file 'thermophysicalProperties'.
// Set initial gas composition in directory '0' (Update O2, N2, files may be added as necessary for other species present at t=0).

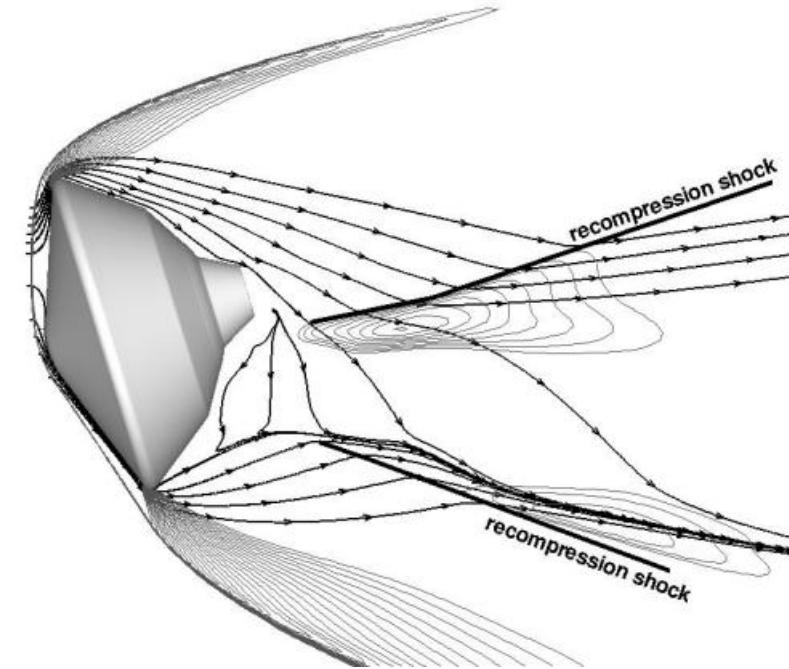
multiComponent no;
// yes: Multicomponent diffusion is used - Average diffusion coefficients are computed with Mutation++ and Fick's law is used
// Only available when mutationTT is switched to 'yes' - defaults to 'multiComponent = no' otherwise.
// Option available in finite-rate chemistry mode only (finiteRate=yes).
// no: Binary diffusion is used - an equal diffusion coefficient D is used for all molecules (update D in 'constant/porousMat/constantProperties')

elementConservation no;
// yes: Element conservation equation is solved in equilibrium mode.
// Reduce the max time step to 1e-4 to prevent instability.
// no: Used when element fraction is constant (e.g. type 1 and 2 models), and in finite-rate chemistry mode (species conservation is automatically used).
```

# Aerothermal environment computed from DPLR\*

## DPLR assumptions

- laminar boundary layer
- chemical and thermal non-equilibrium
- radiative equilibrium
- super-catalytic wall
- non-blowing & smooth wall
- 12 reactions & 8 species
- Mars atmosphere:  $y_{CO_2} = 0.97$ ,  $y_{N_2} = 0.03$

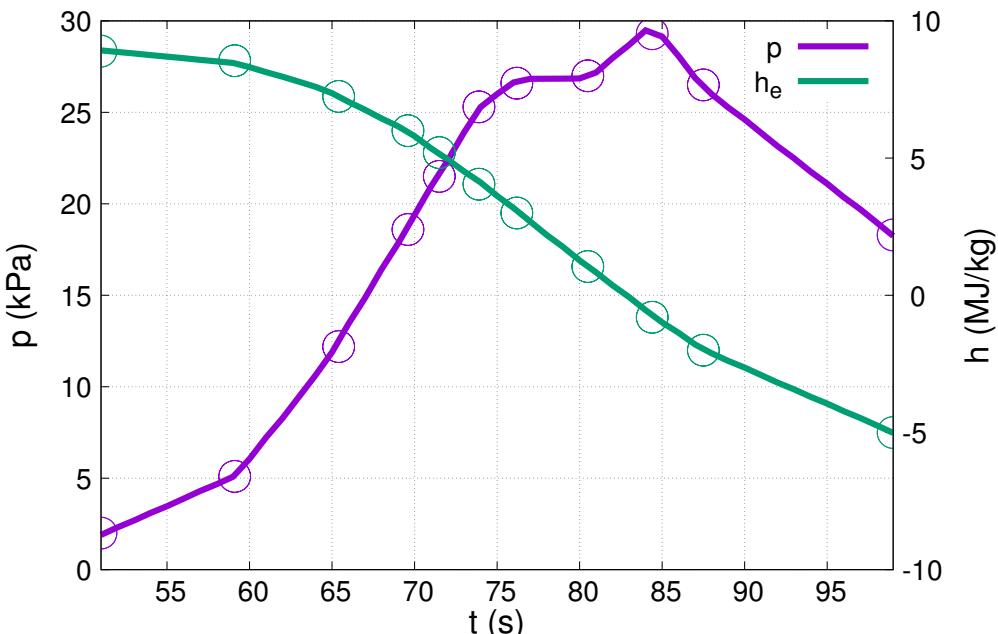


**BLAYER** calculates the **boundary layer edges** using a curvature-based method

**surface pressure  $p_w$ , heat transfer coefficient  $C_H$  and enthalpy  $h_e$**  at the **boundary layer edges** are used as inputs in the **material response code: PATO**

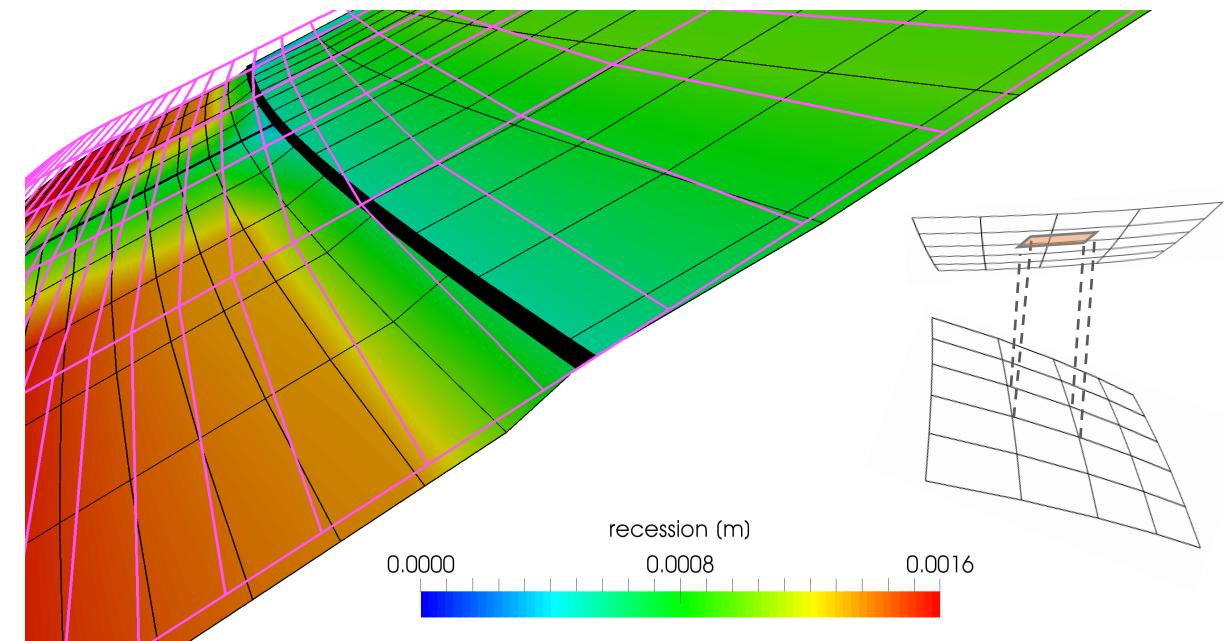
# Temporal and spatial interpolations

## temporal interpolation



11 discrete times  
(50s to 100s of MSL entry)  
linear interpolation

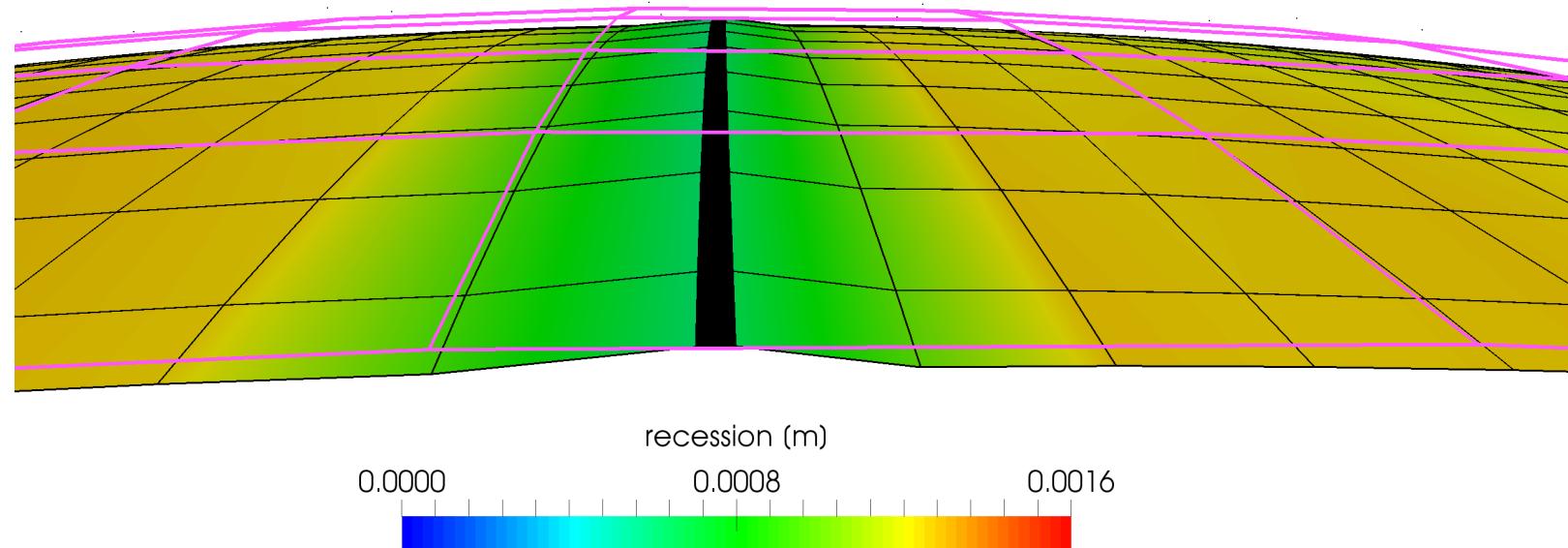
## spatial interpolation



separate mesh regions are numerically **connected** by the **Arbitrary Mesh Interface (AMI)** utility using local **Galerkin projection** implemented in **OpenFOAM**

# “Fencing” effect at tiles interfaces

**MSL heatshield  
front surface  
at the nose**



**Post-test  
arcjet coupons**



# Interpolated inputs from BLAYER to PATO

**surface  
pressure**

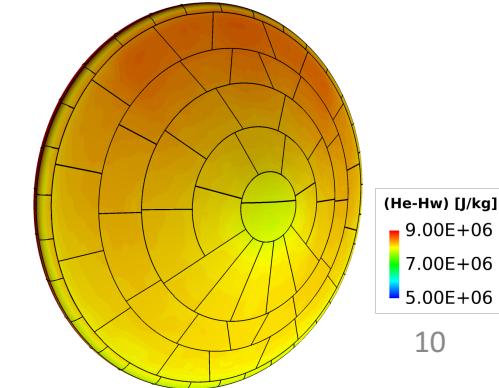
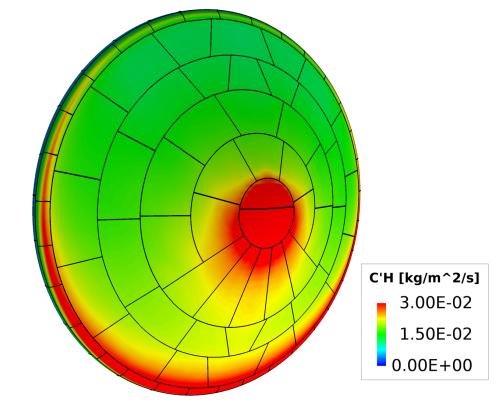
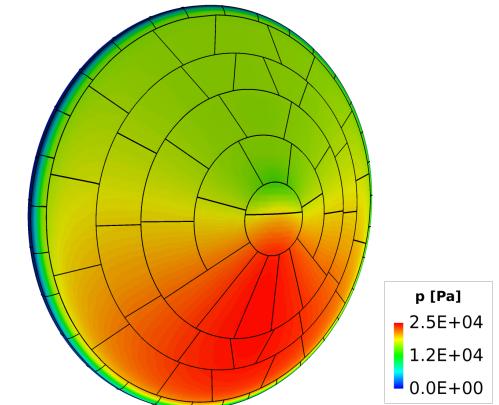
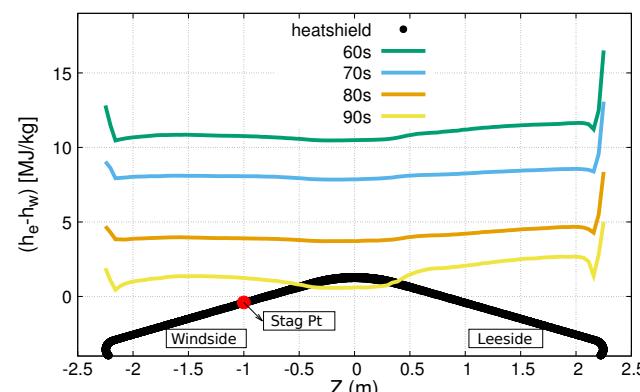
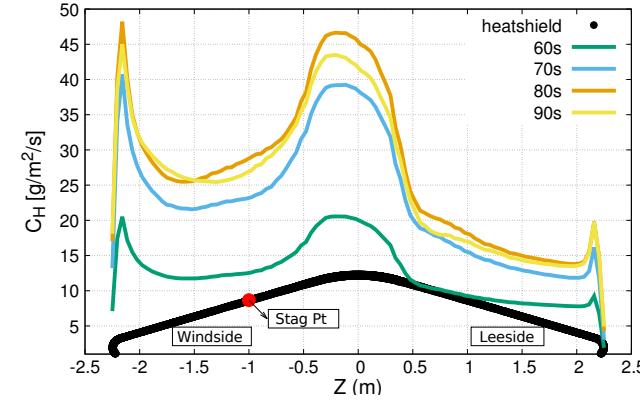
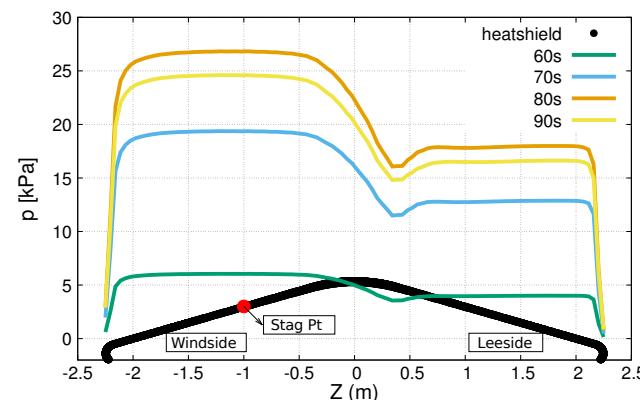
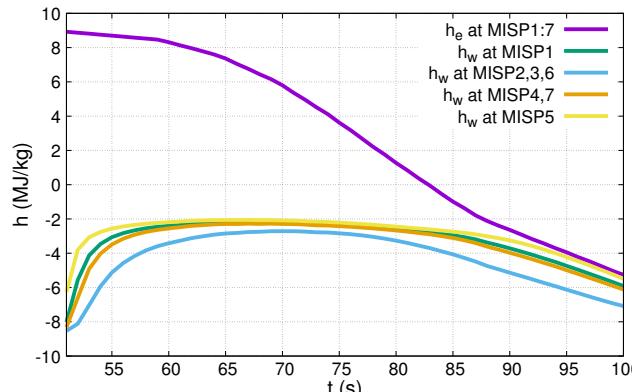
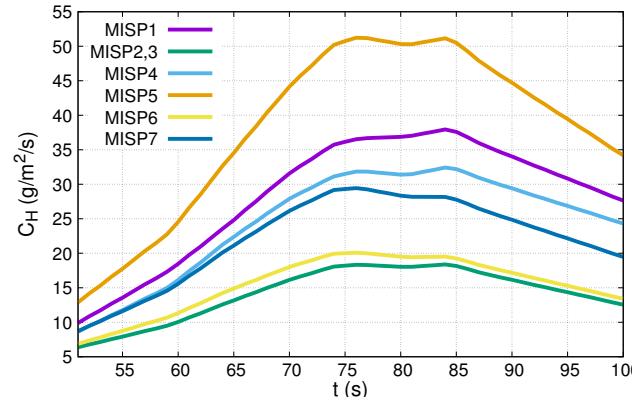
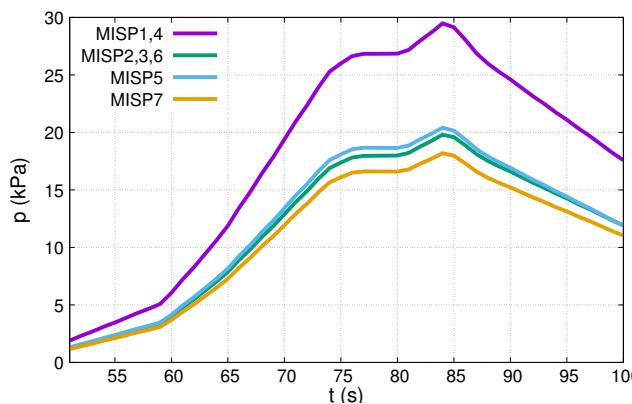
$p_w$

**heat  
transfer  
coefficient**

$C_H$

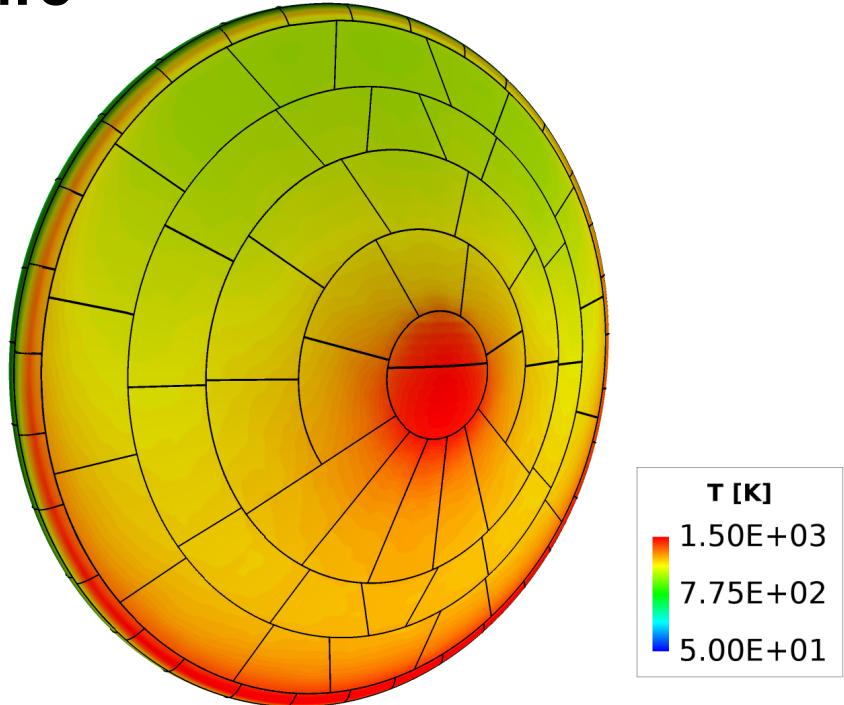
**edge  
enthalpy**

$h_e$

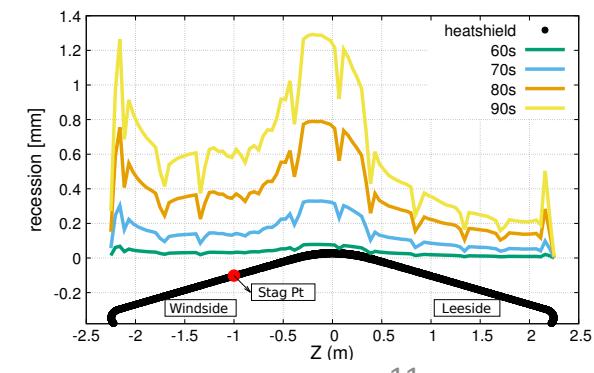
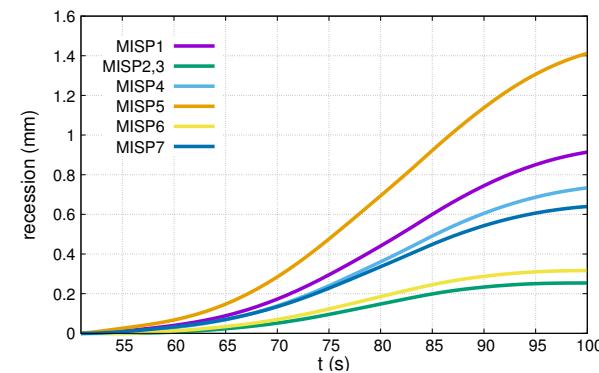
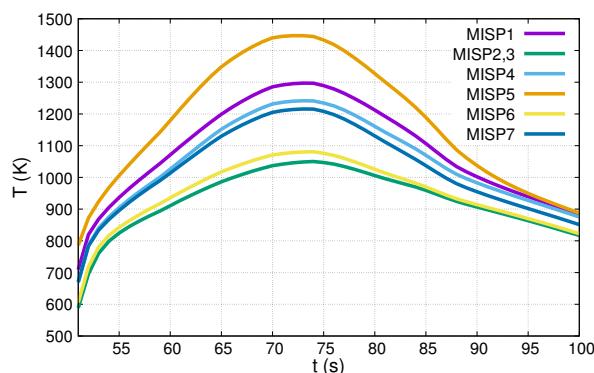
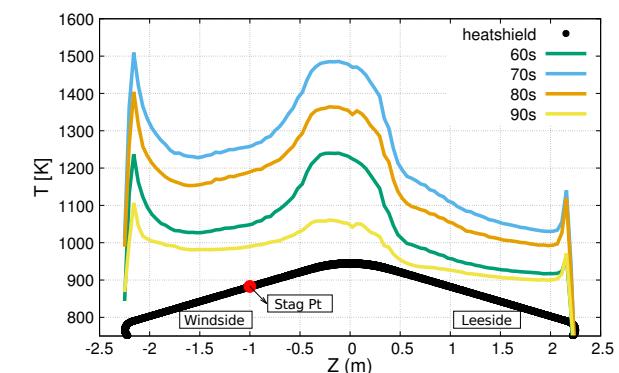
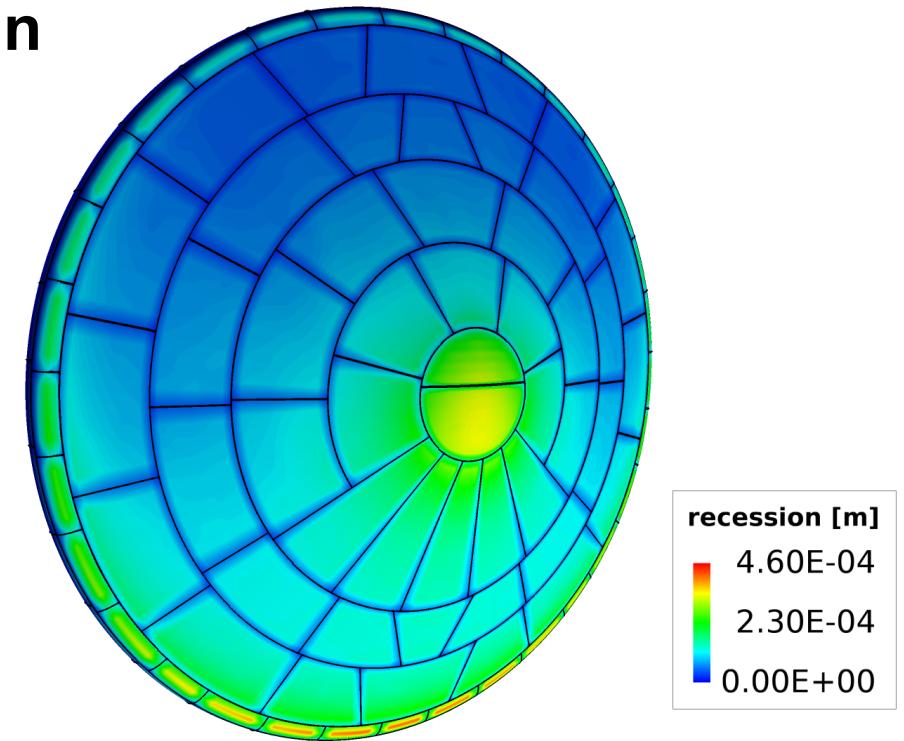


# Temperature and recession from PATO

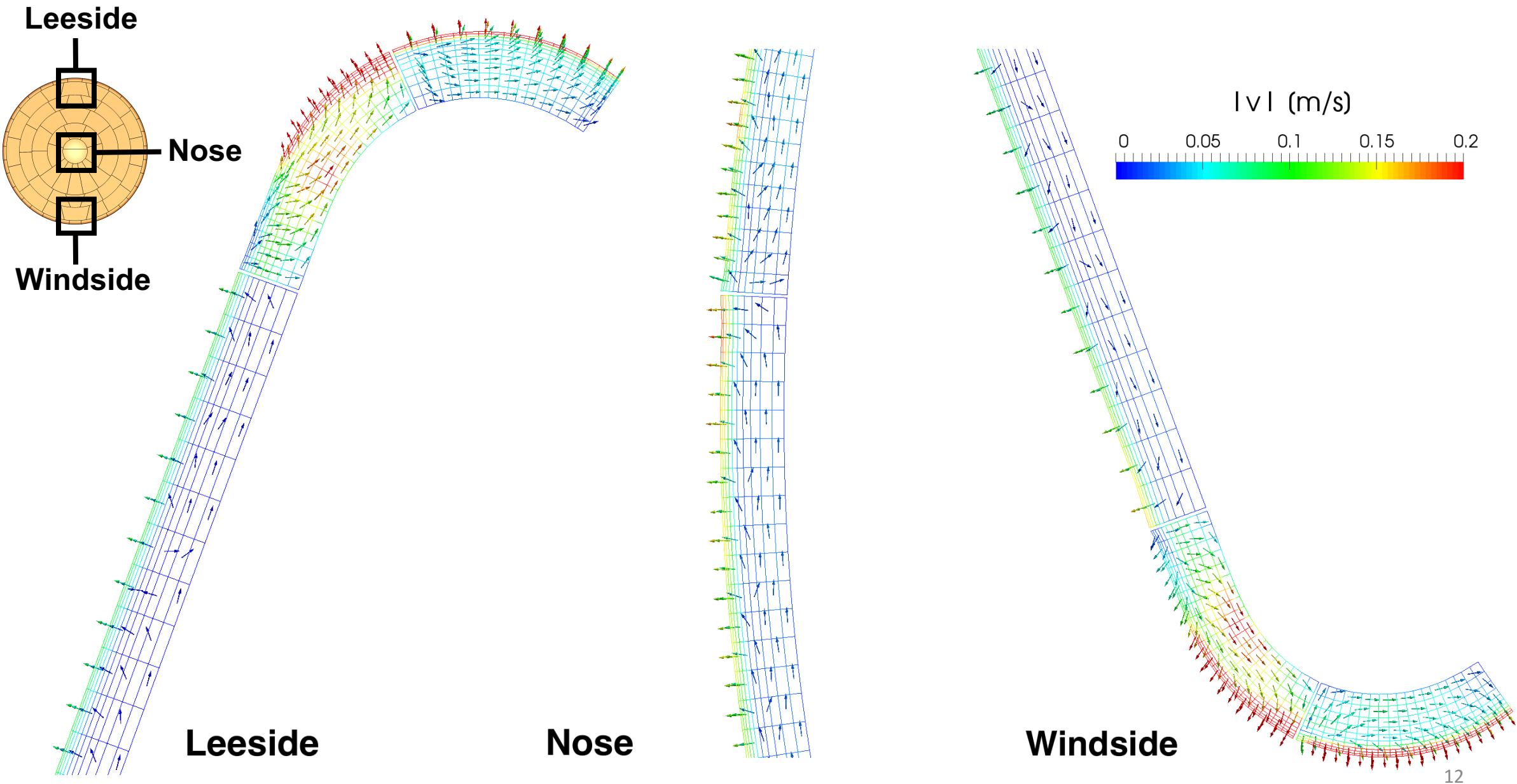
temperature



recession

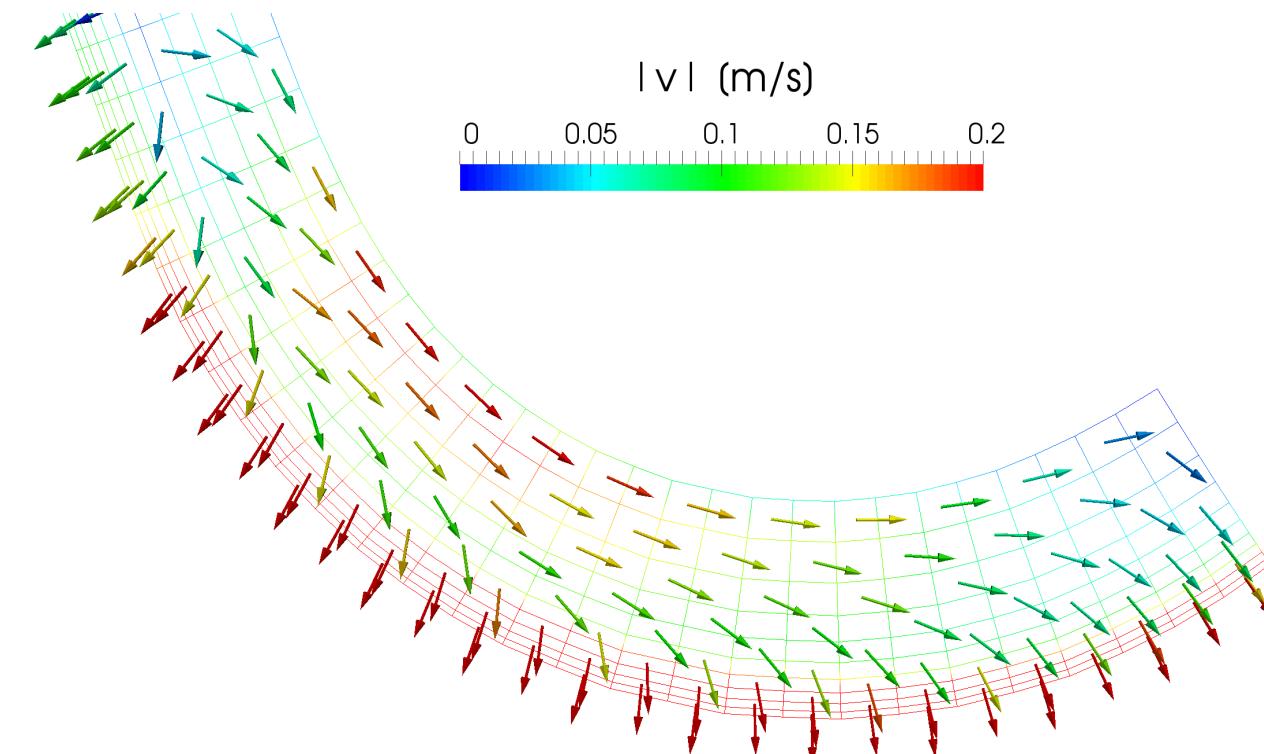


# Velocity inside the porous material

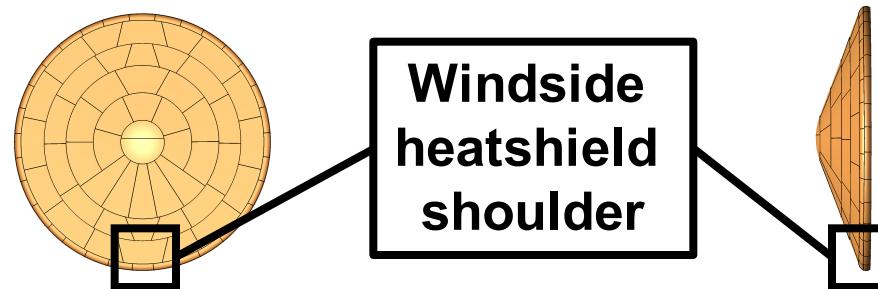
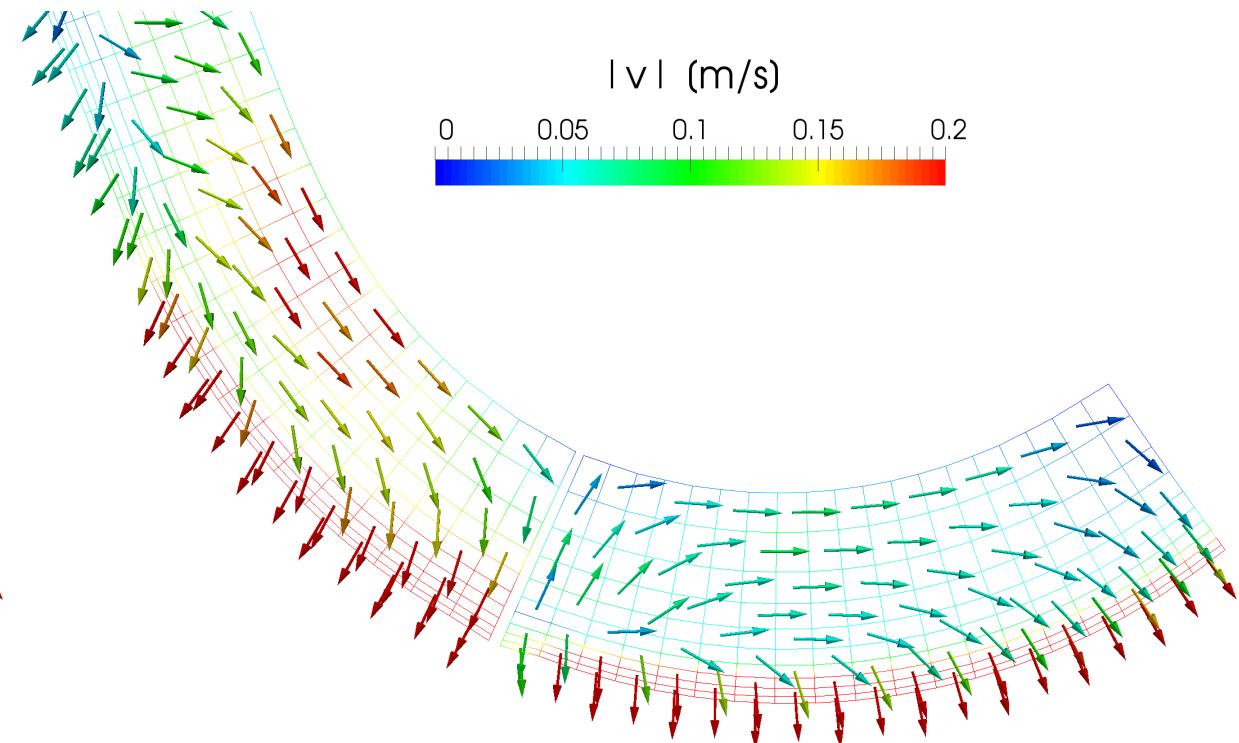


# Tiled configuration changes the flow inside the material

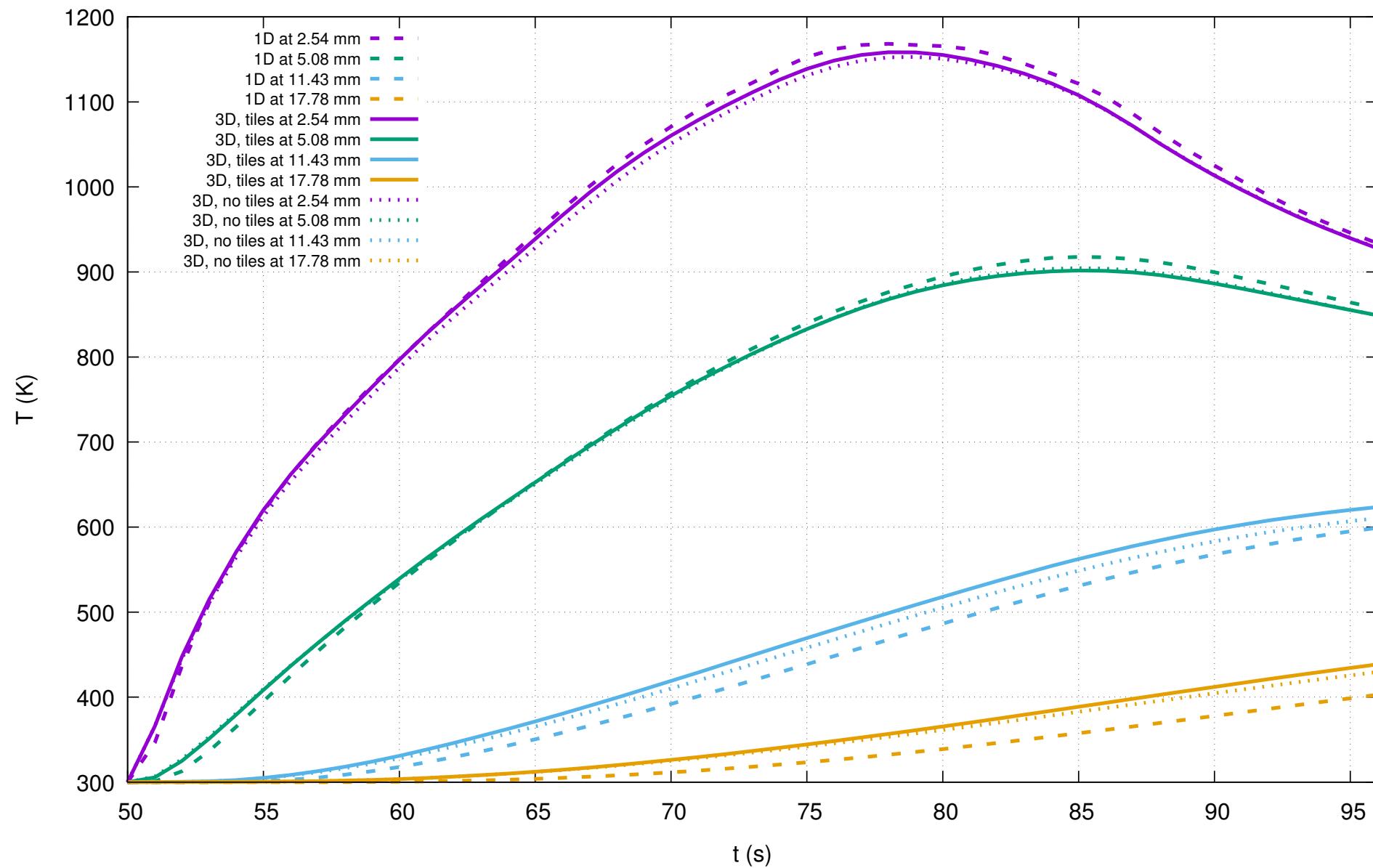
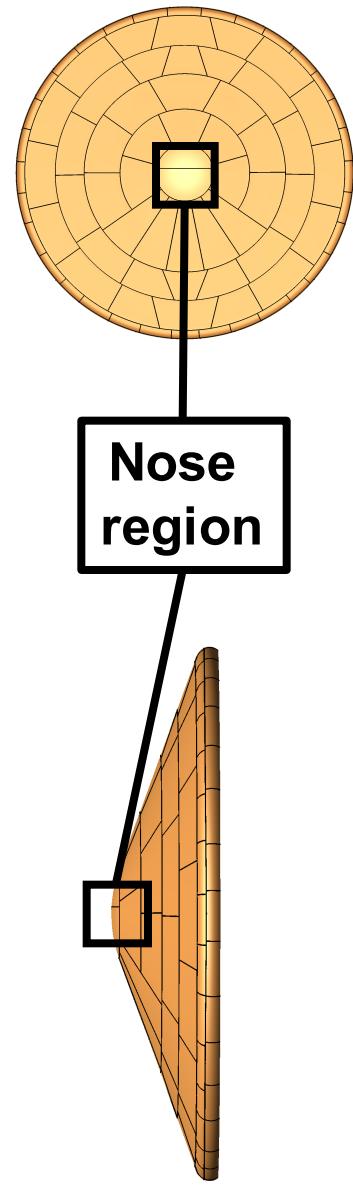
non-tiled configuration



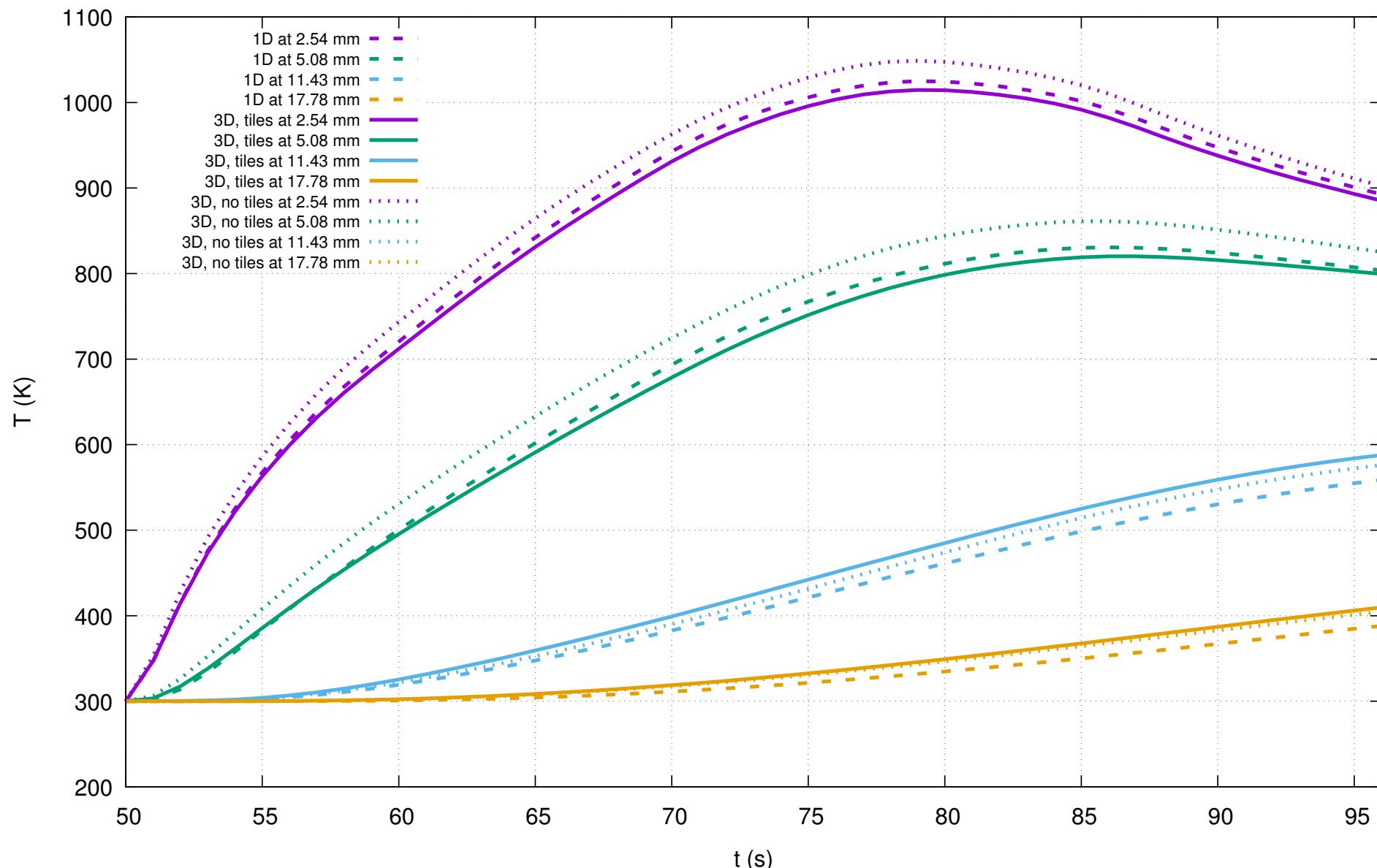
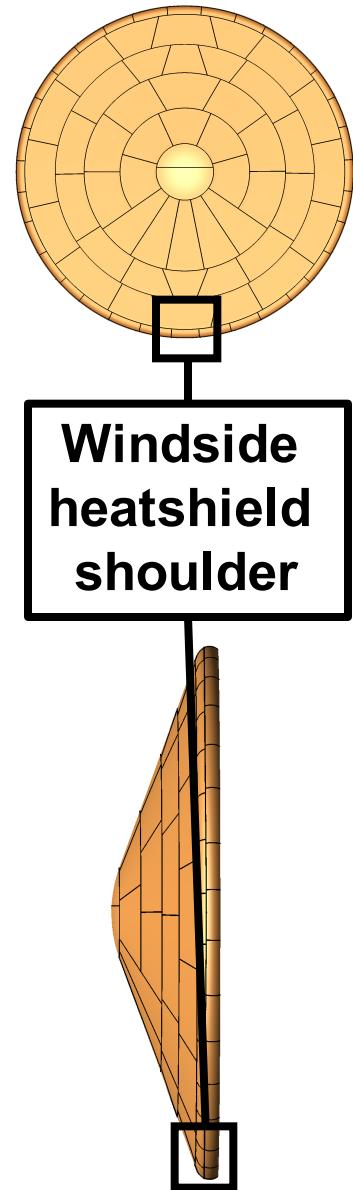
tiled configuration



# 1D and 3D material response comparison – nose



# 1D and 3D material response comparison – shoulder



# Conclusion and future work

## hypersonic environment (DPLR)

- laminar
- super-catalytic wall
- non-blowing
- 8 species & 12 reactions

↓ impacts



## coupling

Linear in time  
Conservative in space by  
local Galerkin projection

Future work includes  
blowing from pyrolysis &  
moving mesh from recession

## porous material response (PATO)

- equilibrium
- pyrolysis
- CMA-type BL approx.
- no finite-rate, no oxidation

↓ outputs

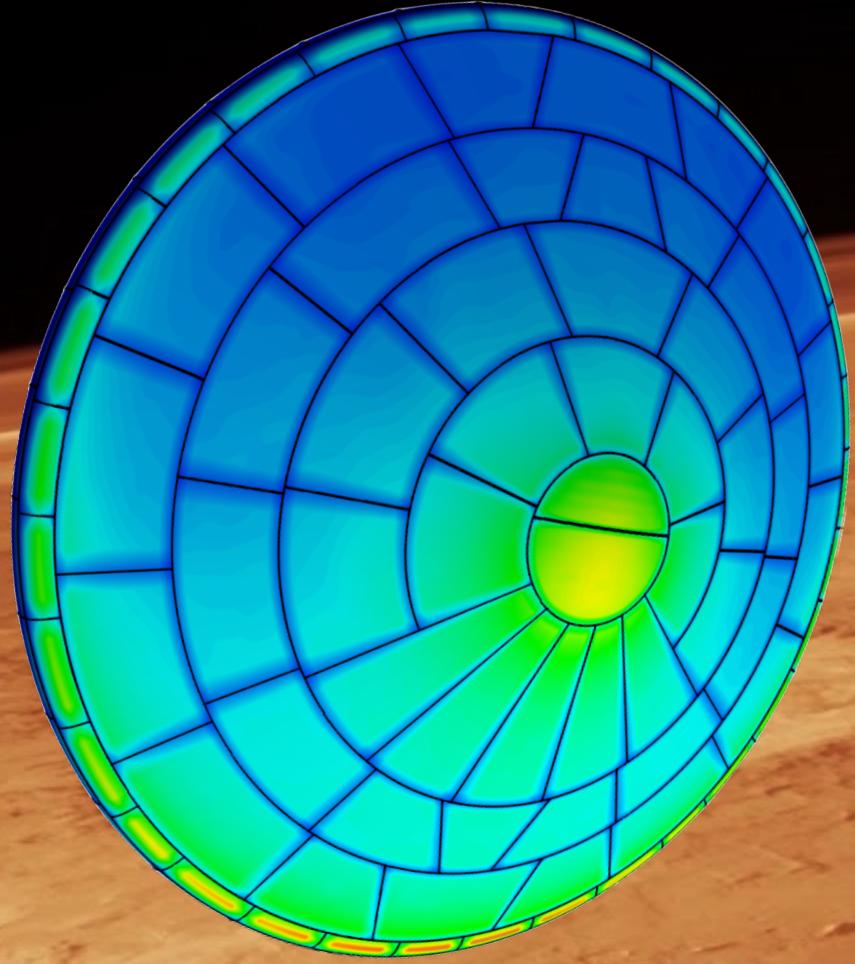
impacts

PATO is a useful tool  
to predict material response  
for future entry missions

- temperature
- recession
- heat fluxes
- inside velocity

# References

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# Questions ?

## 9<sup>th</sup> Ablation Workshop

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### Contact

Jeremie B. E. Meurisse  
650 237 9942  
[jeremie.b.meurisse@nasa.gov](mailto:jeremie.b.meurisse@nasa.gov)